

# Are We Alone in the Universe?

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# Thanks!

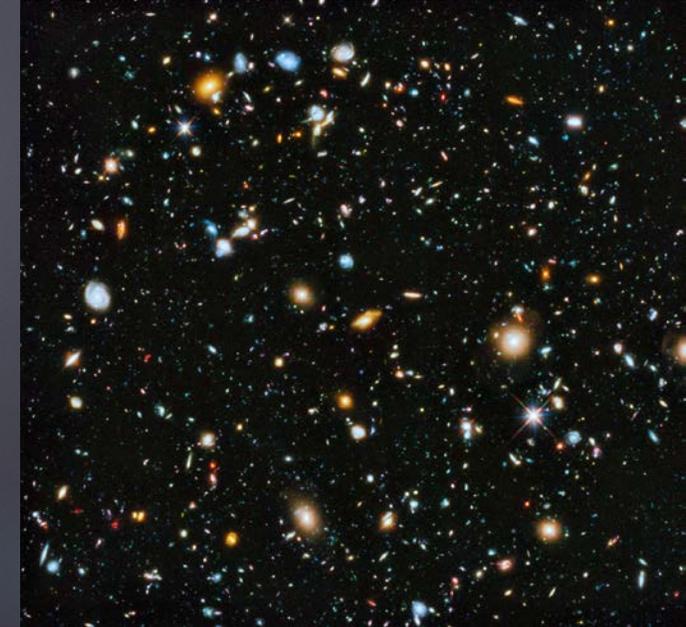
- 1) Honors Carolina
- 2) Morehead-Cain Scholars Program
- 3) UNC Department of Physics and Astronomy

for inviting me to serve as the 2018  
Morehead-Cain Alumni Visiting  
Distinguished Honors Professor

I: LIFE ON EARTH: COULD IT HAPPEN ELSEWHERE?

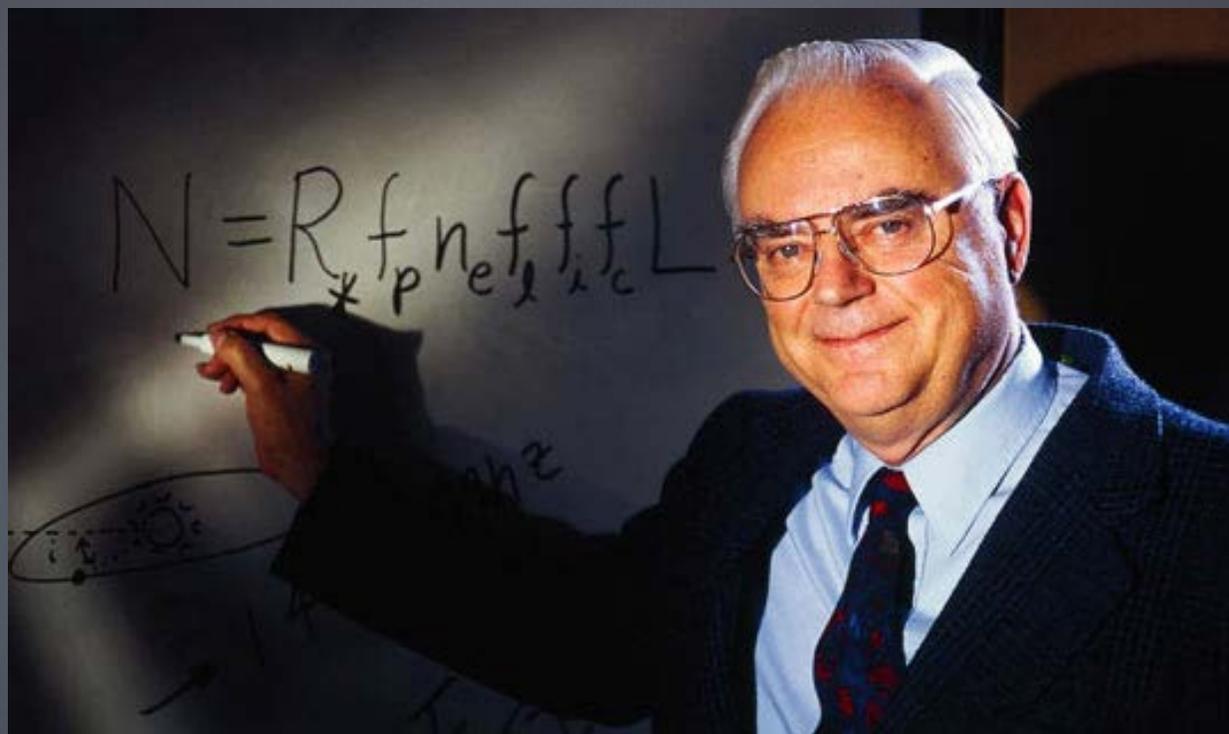
II: MAKING ALIENS

III: TALKING WITH ALIENS



How Many Civilizations are there?

## The Drake Equation



It's not quantitative ... but it is informative ... and can guide our thinking

Let  $N$  = number of communicating civilizations in the galaxy right now

$N$  = number of suitable planets in the galaxy

× chance of a communicating civilization arising on a suitable planet

× survival fraction

$$N = N_{\text{astro}} \times f_{\text{bio}} \times (\text{lifetime term})$$

$$= (N_{\star} \times f_s \times N_p \times f_e) \times (f_L \times f_i \times f_C) \times L/L_{\text{MW}}$$

Investigate each term...

# Drake Equation



×



×



×



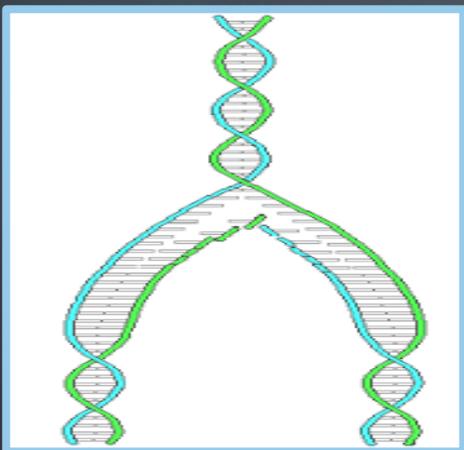
$N_{\star}$

$f_s$

$N_p$

$f_e$

= **N**



×



×



×



$f_L$

$f_i$

$f_c$

$L/L_{MW}$

# Number of Stars in the Galaxy



$N_{\star}$

We actually know this pretty accurately!

$$\begin{aligned} N_{\star} &= 3 \times 10^{11} \\ &= 300,000,000,000 \\ &= 300 \text{ billion} \\ &= 300,000 \text{ million} \end{aligned}$$

$$N_{\star} = 3 \times 10^{11}$$

# Fraction of stars suitable for life



$f_s$

What makes a star suitable?

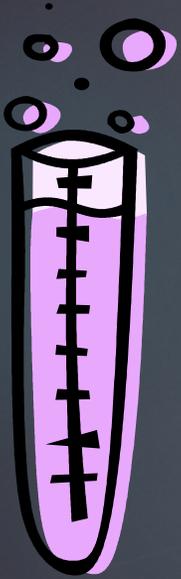
What do we need from the star?

How do stars vary?

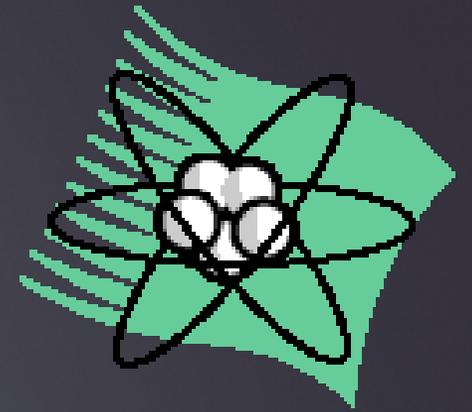
# Energy of the Stars



- ▶ Most energy we are familiar with is *chemical energy*
- ▶ It is released from re-arranging chemical bonds in atoms and molecules
- ▶ This is what powers cars, BBQs, and even rockets (as we will see later)
- ▶ But it does not provide enough energy to power the stars!



# Nuclear Energy



- ▶ The key is Einstein's formula for mass-energy equivalence:

$$E=mc^2$$

- ▶ This relates a change in **mass** to a change in **energy**

# H “burning” in the Sun

The simplest nuclear reactions involve fusing 4 H nuclei (protons!) into one He<sup>4</sup> nucleus (an  $\alpha$  particle)



Mass of a proton =  $m_p = 1.673 \times 10^{-27}$  kg

Mass of a <sup>4</sup>He =  $m_\alpha = 6.645 \times 10^{-27}$  kg

Difference in mass = “mass defect”

$$= 4 m_p - m_\alpha$$

$$= 4.7 \times 10^{-29} \text{ kg}$$

**Mass is lost each time!**

# H "burning" in the Sun

- ▶ Each reaction releases an energy  $E = mc^2$
- ▶ So in our case that is

$$\begin{aligned} E &= (4.7 \times 10^{-29}) \times (3.0 \times 10^8)^2 \text{ kg m}^2/\text{s}^2 \\ &= 4.3 \times 10^{-12} \text{ Joule} \\ &= \text{not much} \end{aligned}$$

How far can we lift it?



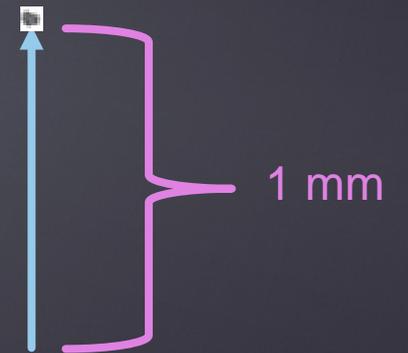
Poppy seeds....



Take one....



Cut into 1000 pieces....



Take one piece...

# But there are lots of reactions!

- ▶ We can calculate how many of these reactions occur
- ▶ The Sun emits  $3.90 \times 10^{26}$  Joule/s
- ▶ So that is

$3.90 \times 10^{26} / 4.3 \times 10^{-12}$  reactions per second

$= 9 \times 10^{37}$  reactions per second

That is a **HUGE** number!



# How much mass is this?

- ▶ There are  $9 \times 10^{37}$  reactions per second
- ▶ Each reaction destroys  $4.7 \times 10^{-29}$ kg
- ▶ So the total mass destroyed each second is:

$$(9 \times 10^{37}) \times (4.7 \times 10^{-29}\text{kg}) = 4.2 \times 10^9 \text{ kg}$$

**That's 4.2 million tonnes per second!**

# How much H burns?

- ▶ Since we now know how many reactions there are per second...  
...we can work out how much H is burned each second.

There are  $9 \times 10^{37}$  reactions

Each consumes 4 H nuclei

$$\begin{aligned}\text{So we burn } 4 \times m_p \times 9 \times 10^{37} &= 4 \times 1.673 \times 10^{-27} \text{ kg} \times 9 \times 10^{37} \\ &= 6.02 \times 10^{11} \text{ kg} \\ &= 602 \text{ million tonnes per second!}\end{aligned}$$

# How much He is produced?

- ▶ Since we now know how many reactions there are per second...  
...we can work out how much He is produced each second.

There are  $9 \times 10^{37}$  reactions

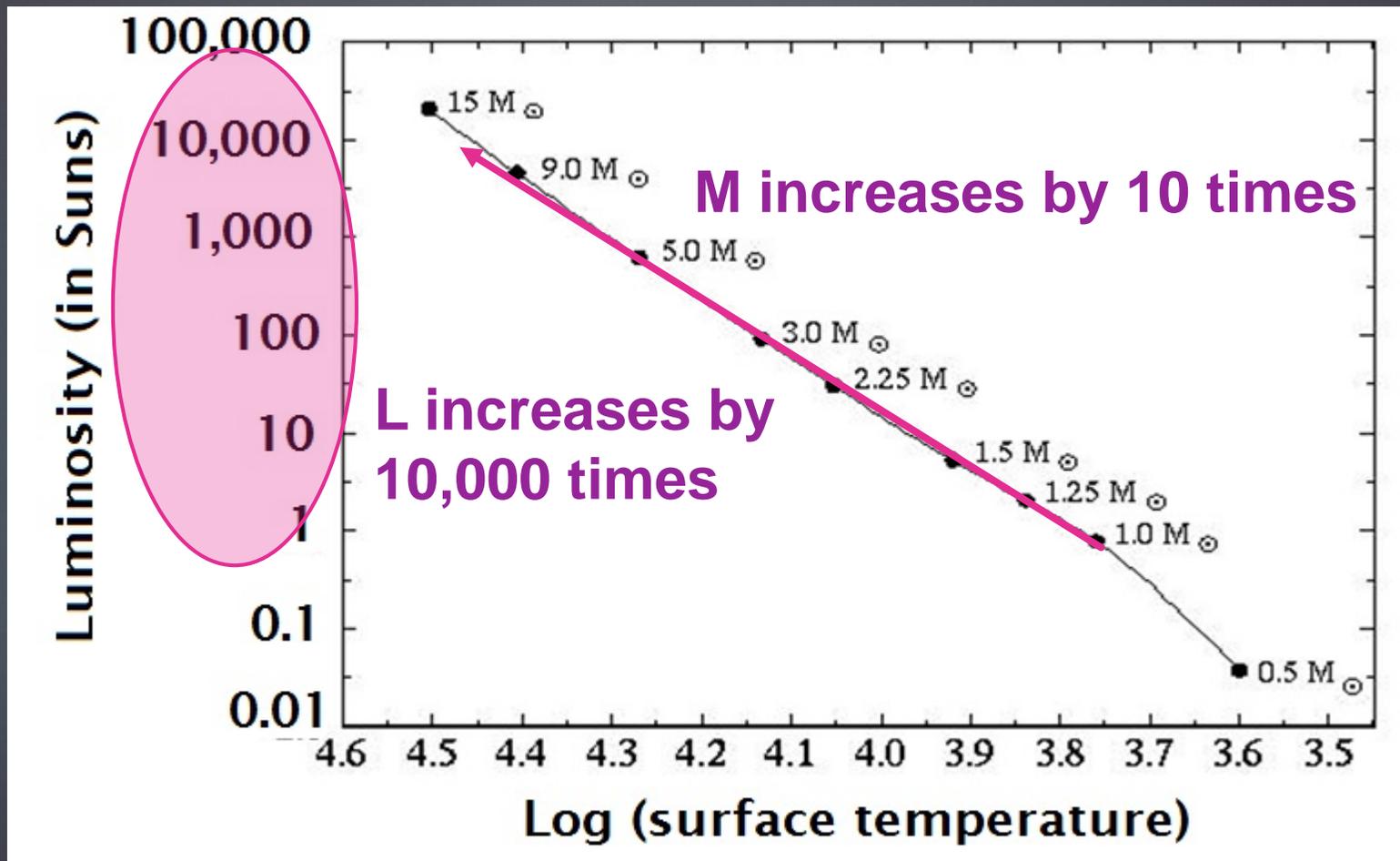
Each produces one  ${}^4\text{He}$  nucleus

$$\begin{aligned}\text{So we produce } m_{\alpha} \times 9 \times 10^{37} &= 6.645 \times 10^{-27} \text{ kg} \times 9 \times 10^{37} \\ &= 5.98 \times 10^{11} \text{ kg} \\ &= \mathbf{598 \text{ million tonnes per second!}}\end{aligned}$$

And as we might hope – the difference between these is 4 million tonnes as we calculated initially!

# The Main Sequence

- ▶ By making models for different masses we have learned that the **Main Sequence** is a sequence of stars of **different mass**



# Mass-Luminosity Relation

- ▶ One can show that roughly we can relate mass  $M$  to luminosity  $L$  by:  $L \propto M^3$

- ▶ So to make this an exact equation we insert a constant  $k$  which we have to determine:

$$L = kM^3$$

- ▶ If we measure mass and luminosity in solar units then  $k=1$  because the Sun has a mass of 1 and a luminosity of 1 😊

- ▶ The symbol for the Sun is  $\odot$ , so we can write the mass-luminosity relation as

$$\frac{L}{L_{\odot}} = \left( \frac{M}{M_{\odot}} \right)^3$$

# How does this affect lifetimes of stars?

- ▶ Stars live while they have fuel to burn
- ▶ Massive stars burn brightly and use fuel quickly
- ▶ Massive stars do not live very long...



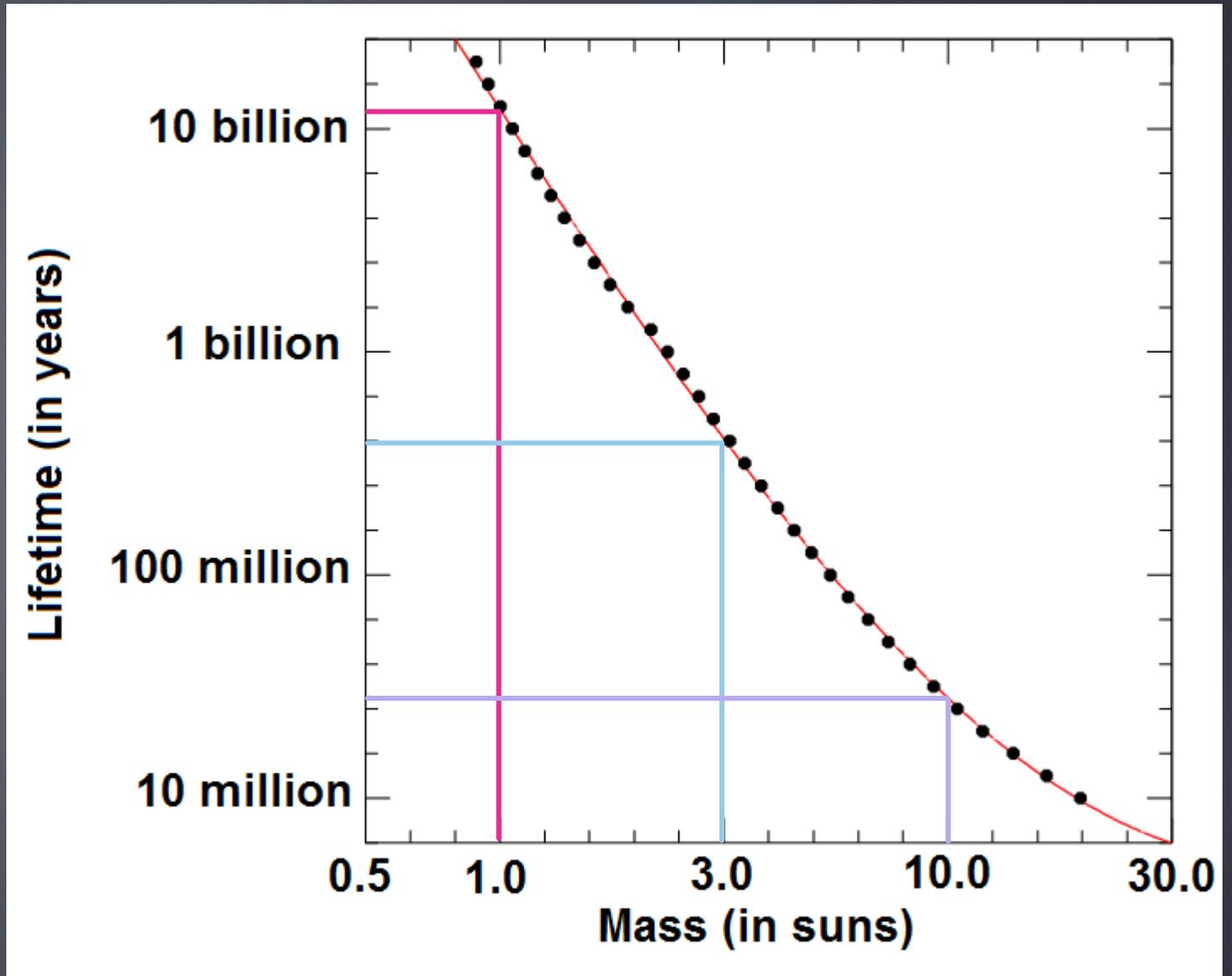
- The Sun will live for about 10 *billion* years
- A 10 solar mass star lives for about 30 *million* years

# Stellar Ages

Sun: 10 billion years

M = 3 times Sun: 400 million years

M = 10 times Sun: 30 million years



# Mass-age relation

- ▶ We can use the mass-luminosity relation to derive a mass-age relation
- ▶ The amount of fuel available is just the mass of the star.
- ▶ So the lifetime is the rate of using fuel (**the luminosity**) divided by the amount of fuel available (**the mass**)
- ▶ Lifetime  $t$  is proportional to  $M/L$
- ▶ Of course the star cannot use all of its mass...it can only burn where it is hot (in the middle)
- ▶ So we put in a constant of proportionality again

$$t = k \left( \frac{M}{L} \right) = k \left( \frac{M}{M^3} \right) = \frac{k}{M^2}$$

# Mass-age relation

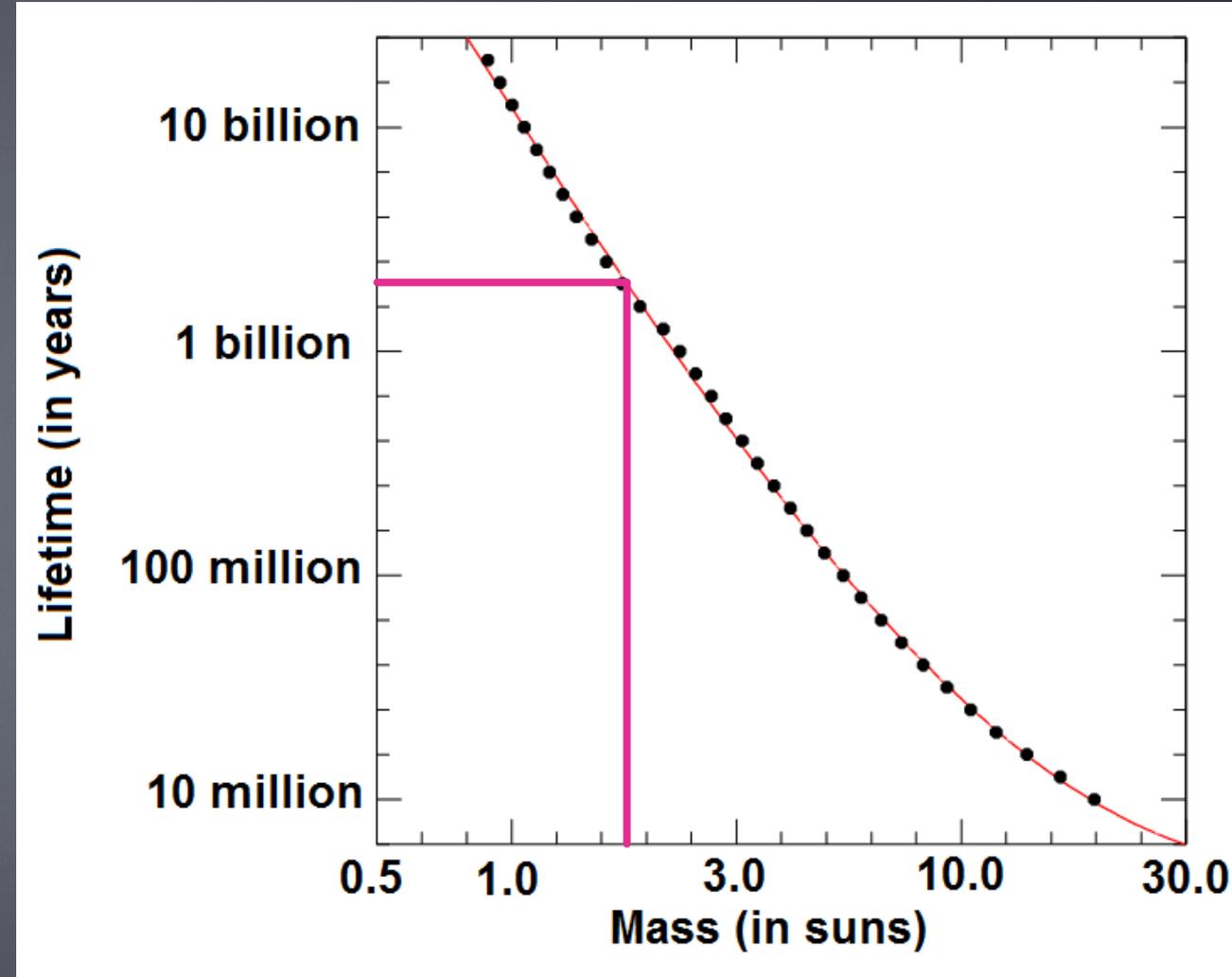
- ▶ Lets again measure things in solar units.
- ▶ The lifetime of the Sun is  $t_{\odot} = 10^{10}$  years
- ▶ So for a star of a different mass  $M$  we get

$$\frac{t}{t_{\odot}} = \left( \frac{M}{M_{\odot}} \right)^{-2}$$

These relations are *approximate*. The values used in lectures are the result of very detailed computer calculations. But the essence is in these simple equations!

# How long do we need?

- ▶ How long does it take to produce a communicating civilization?
- ▶ Clearly we need the star to live at least that long...
- ▶ Earth?
  - 0.5 billion years for first life
  - 4 billion years for large life
  - 4.5 billion years for **us** to develop communication technology



Maybe assume 2 billion years as a minimum?

This gives a maximum mass of about 2 Suns.