

Are We Alone in the Universe?

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Thanks!

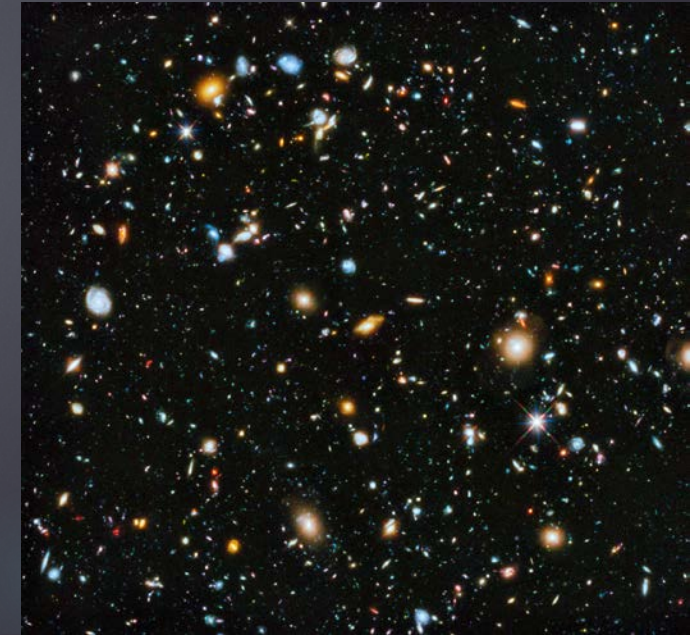
- 1) Honors Carolina
- 2) Morehead-Cain Scholars Program
- 3) UNC Department of Physics and Astronomy

for inviting me to serve as the 2018
Morehead-Cain Alumni Visiting
Distinguished Honors Professor

I: LIFE ON EARTH: COULD IT HAPPEN ELSEWHERE?

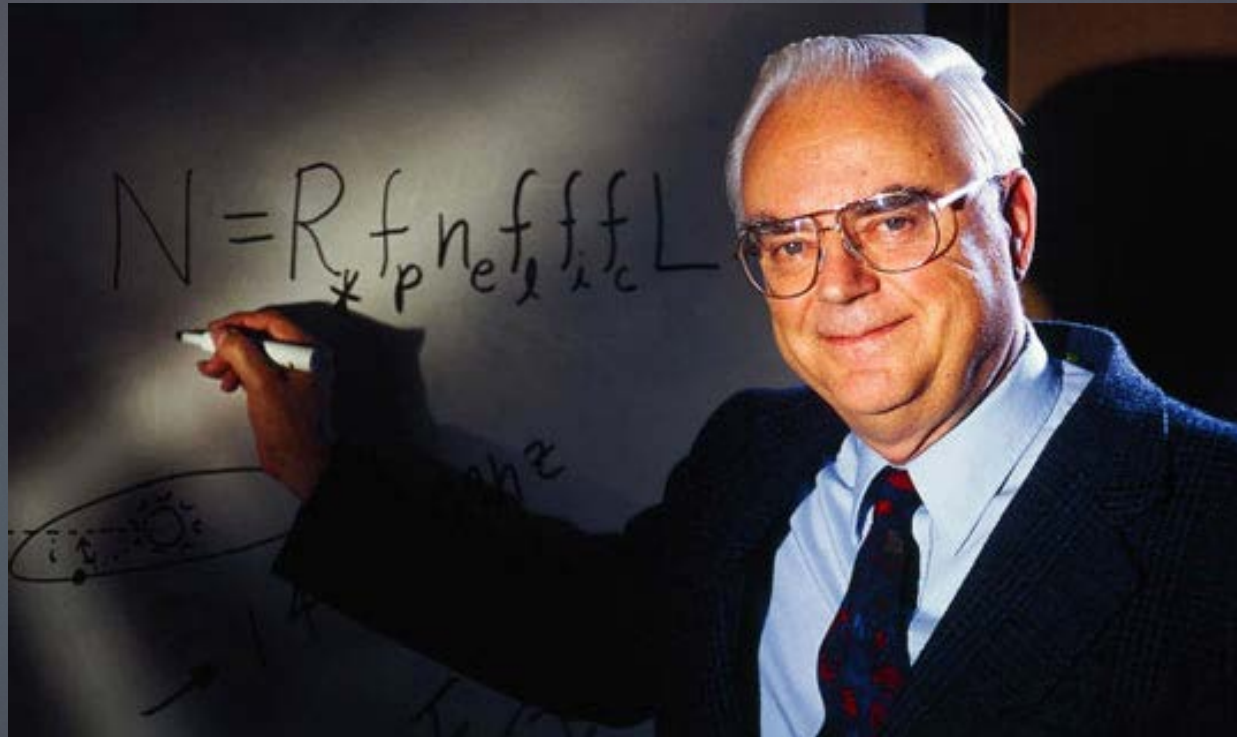
II: MAKING ALIENS

III: TALKING WITH ALIENS



How Many Civilizations are there?

The Drake Equation



It's not quantitative ... but it is informative ... and can guide our thinking

Let N = number of communicating civilizations in the galaxy right now

N = number of suitable planets in the galaxy

× chance of a communicating civilization arising on a suitable planet

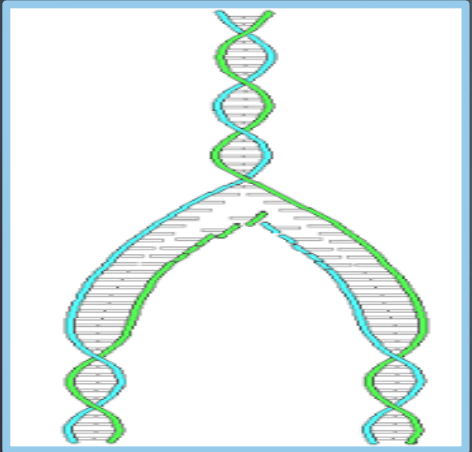
× survival fraction

$$N = N_{\text{astro}} \times f_{\text{bio}} \times (\text{lifetime term})$$

$$= (N_{\star} \times f_s \times N_p \times f_e) \times (f_L \times f_i \times f_c) \times L/L_{\text{MW}}$$

Investigate each term...

Drake Equation

 N_{\star} \times  f_s \times  N_p \times  f_e $=$ N  f_L \times  f_i \times  f_c \times  L/L_{MW}

Number of Stars in the Galaxy



N_{\star}

We actually know this pretty accurately!

$$\begin{aligned} N_{\star} &= 3 \times 10^{11} \\ &= 300,000,000,000 \\ &= 300 \text{ billion} \\ &= 300,000 \text{ million} \end{aligned}$$

$$N_{\star} = 3 \times 10^{11}$$

Fraction of stars suitable for life



f_s

What makes a star suitable?

What do we need from the star?

How do stars vary?

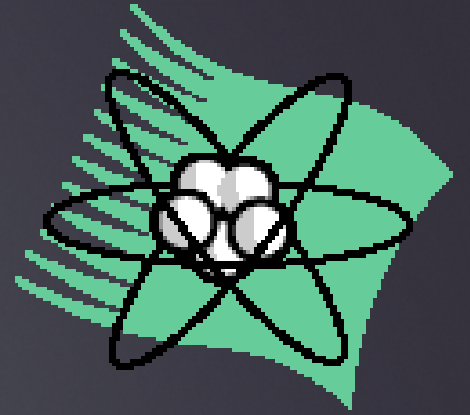
Energy of the Stars



- ▶ Most energy we are familiar with is *chemical energy*
- ▶ It is released from re-arranging chemical bonds in atoms and molecules
- ▶ This is what powers cars, BBQs, and even rockets (as we will see later)
- ▶ But it does not provide enough energy to power the stars!



Nuclear Energy



- ▶ The key is Einstein's formula for mass-energy equivalence:

$$E=mc^2$$

- ▶ This relates a change in mass to a change in energy

H “burning” in the Sun

The simplest nuclear reactions involve fusing 4 H nuclei (protons!) into one He^4 nucleus (an α particle)



Mass of a proton = $m_p = 1.673 \times 10^{-27} \text{ kg}$

Mass of a ${}^4\text{He}$ = $m_\alpha = 6.645 \times 10^{-27} \text{ kg}$

Difference in mass = “mass defect”

$$= 4 m_p - m_\alpha$$

$$= 4.7 \times 10^{-29} \text{ kg}$$

Mass is lost each time!

H “burning” in the Sun

- ▶ Each reaction releases an energy $E = mc^2$
- ▶ So in our case that is

$$\begin{aligned} E &= (4.7 \times 10^{-29}) \times (3.0 \times 10^8)^2 \text{ kg m}^2/\text{s}^2 \\ &= 4.3 \times 10^{-12} \text{ Joule} \\ &= \text{not much} \end{aligned}$$



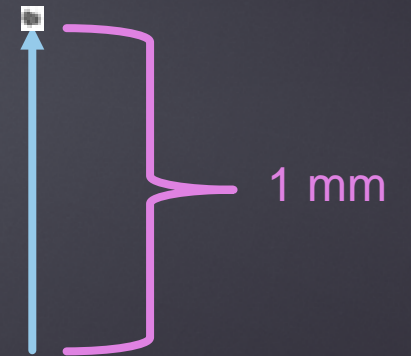
Poppy seeds....



Take one....



Cut into 1000 pieces....



Take one piece...

But there are lots of reactions!

- ▶ We can calculate how many of these reactions occur
- ▶ The Sun emits 3.90×10^{26} Joule/s
- ▶ So that is

$3.90 \times 10^{26} / 4.3 \times 10^{-12}$ reactions per second

$= 9 \times 10^{37}$ reactions per second

That is a **HUGE** number!



How much mass is this?

- ▶ There are 9×10^{37} reactions per second
- ▶ Each reaction destroys $4.7 \times 10^{-29}\text{kg}$
- ▶ So the total mass destroyed each second is:
$$(9 \times 10^{37}) \times (4.7 \times 10^{-29}\text{kg}) = 4.2 \times 10^9 \text{ kg}$$

That's 4.2 million tonnes per second!

How much H burns?

- Since we now know how many reactions there are per second...
...we can work out how much H is burned each second.

There are 9×10^{37} reactions

Each consumes 4 H nuclei

$$\begin{aligned}\text{So we burn } 4 \times m_p \times 9 \times 10^{37} &= 4 \times 1.673 \times 10^{-27} \text{ kg} \times 9 \times 10^{37} \\ &= 6.02 \times 10^{11} \text{ kg} \\ &= 602 \text{ million tonnes per second!}\end{aligned}$$

How much He is produced?

- Since we now know how many reactions there are per second...
...we can work out how much He is produced each second.

There are 9×10^{37} reactions

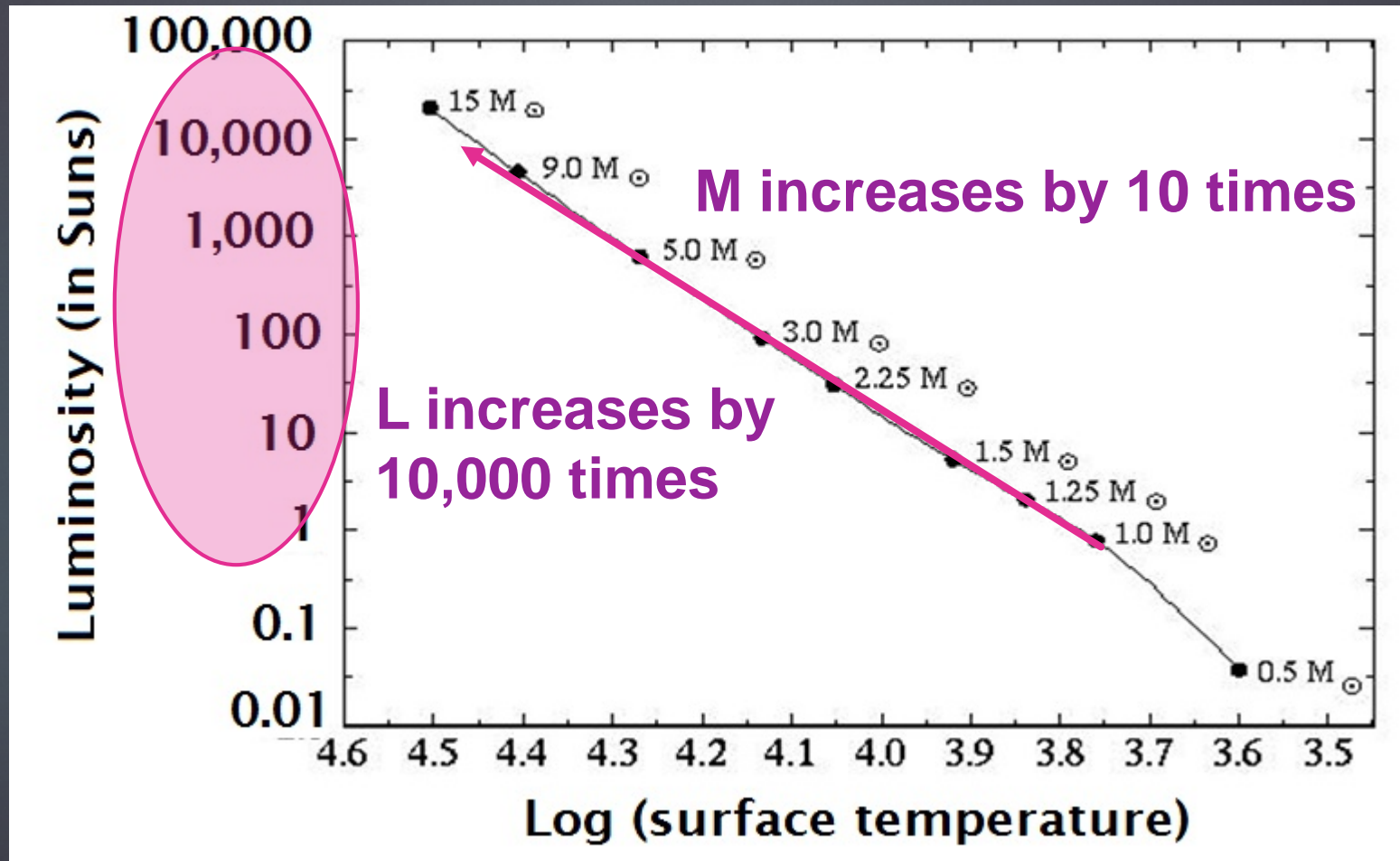
Each produces one ^4He nucleus

$$\begin{aligned}\text{So we produce } m_{\alpha} \times 9 \times 10^{37} &= 6.645 \times 10^{-27} \text{ kg} \times 9 \times 10^{37} \\ &= 5.98 \times 10^{11} \text{ kg} \\ &= \text{598 million tonnes per second!}\end{aligned}$$

And as we might hope – the difference between these is 4 million tonnes as we calculated initially!

The Main Sequence

- By making models for different masses we have learned that the **Main Sequence** is a sequence of stars of **different mass**



Mass-Luminosity Relation

- ▶ One can show that roughly we can relate mass M to luminosity L by: $L \propto M^3$

- ▶ So to make this an exact equation we insert a constant k which we have to determine:

$$L = kM^3$$

- ▶ If we measure mass and luminosity in solar units then $k=1$ because the Sun has a mass of 1 and a luminosity of 1 😊

- ▶ The symbol for the Sun is \odot , so we can write the mass-luminosity relation as

$$\frac{L}{L_{\odot}} = \left(\frac{M}{M_{\odot}} \right)^3$$

How does this affect lifetimes of stars?

- ▶ Stars live while they have **fuel** to burn
- ▶ Massive stars burn **brightly** and use fuel quickly
- ▶ Massive stars do not live very long...



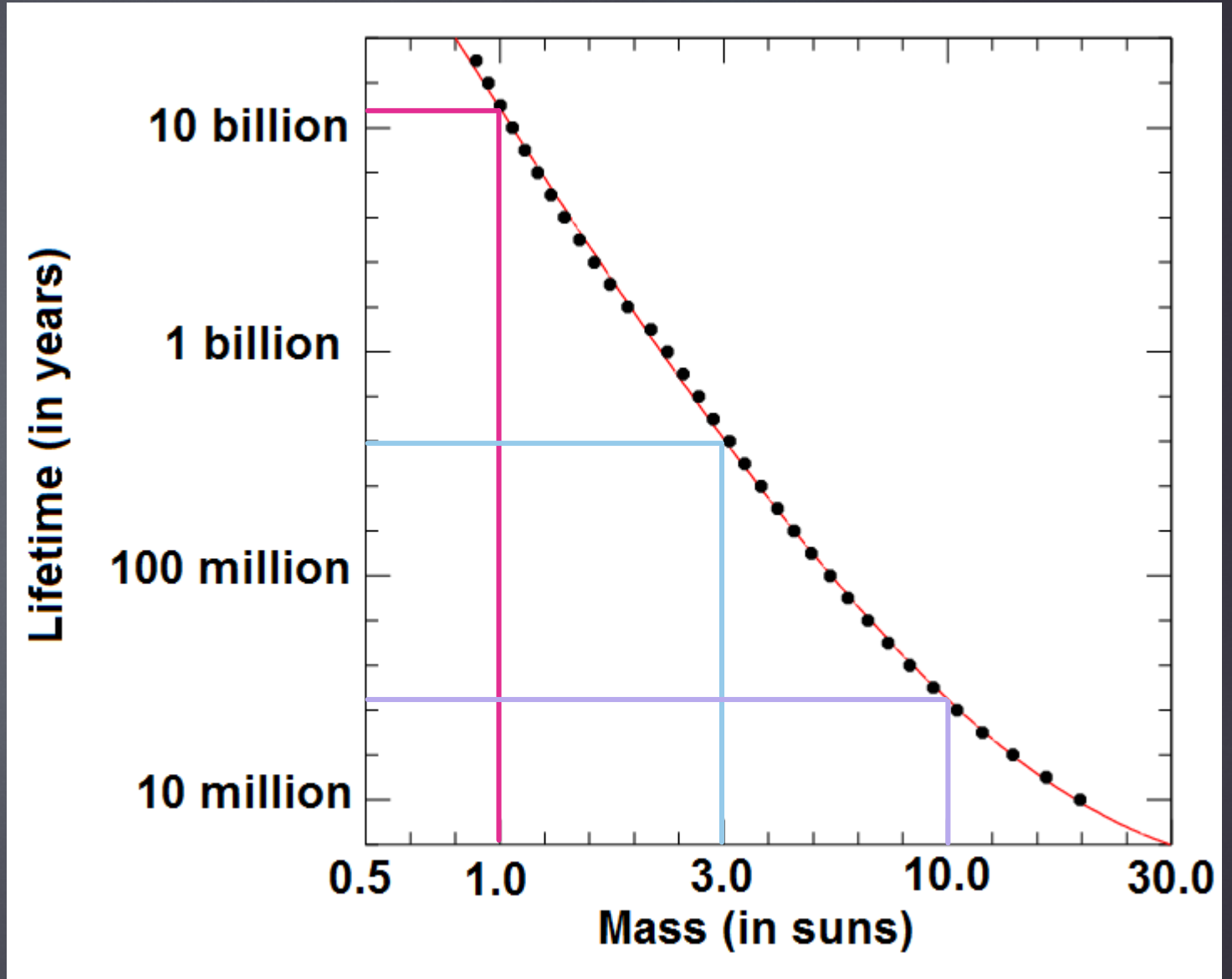
- The Sun will live for about 10 *billion* years
- A 10 solar mass star lives for about 30 *million* years

Stellar Ages

Sun: 10 billion years

$M = 3$ times Sun: 400 million years

$M = 10$ times Sun: 30 million years



Mass-age relation

- ▶ We can use the mass-luminosity relation to derive a mass-age relation
- ▶ The amount of fuel available is just the mass of the star.
- ▶ So the lifetime is the rate of using fuel (the luminosity) divided by the amount of fuel available (the mass)
- ▶ Lifetime t is proportional to M/L
- ▶ Of course the star cannot use all of its mass...it can only burn where it is hot (in the middle)
- ▶ So we put in a constant of proportionality again

$$t = k \left(\frac{M}{L} \right) = k \left(\frac{M}{M^3} \right) = \frac{k}{M^2}$$

Mass-age relation

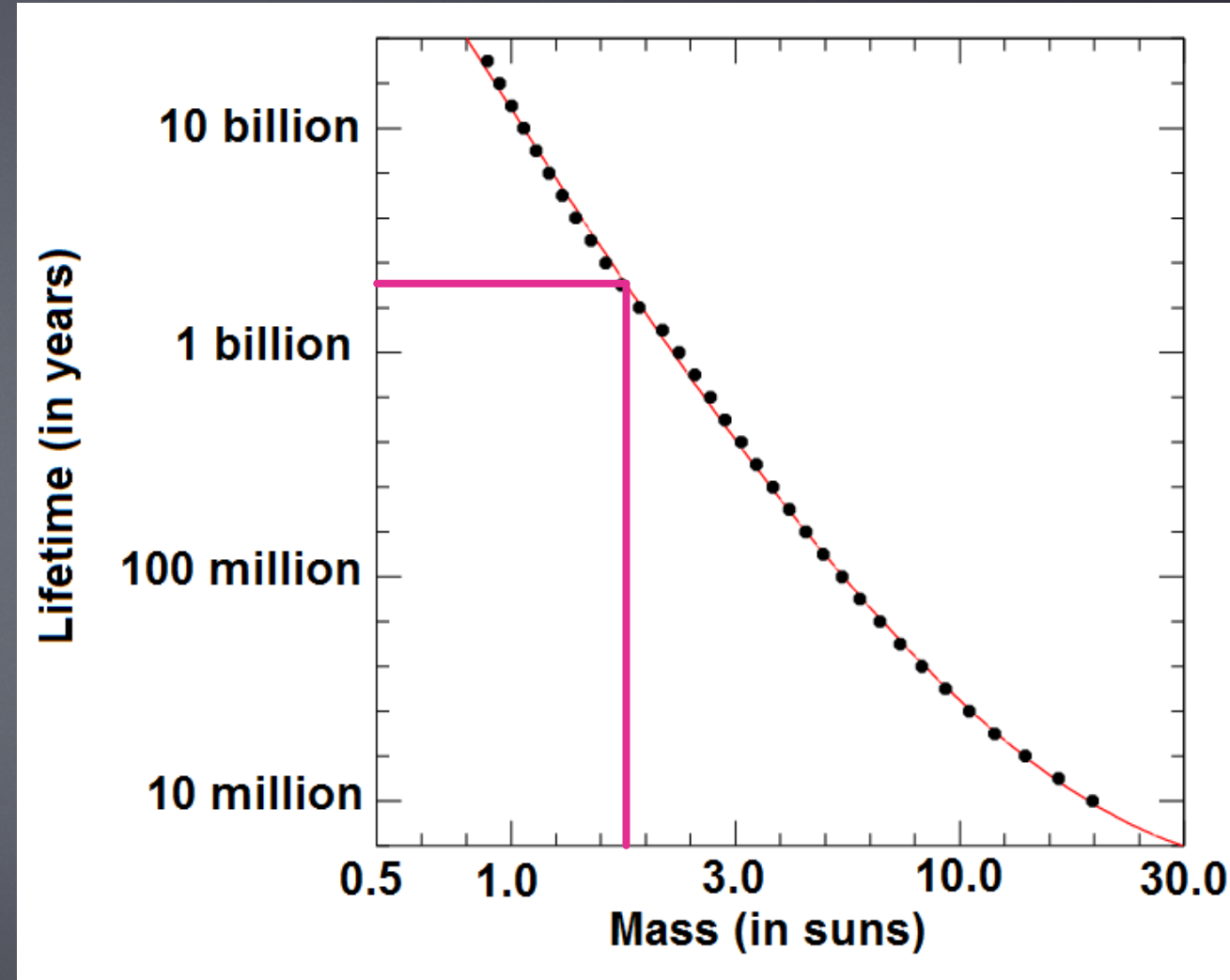
- ▶ Lets again measure things in solar units.
- ▶ The lifetime of the Sun is $t_{\odot} = 10^{10}$ years
- ▶ So for a star of a different mass M we get

$$\frac{t}{t_{\odot}} = \left(\frac{M}{M_{\odot}} \right)^{-2}$$

These relations are *approximate*. The values used in lectures are the result of very detailed computer calculations. But the essence is in these simple equations!

How long do we need?

- ▶ How long does it take to produce a communicating civilization?
- ▶ Clearly we need the star to live at least that long...
- ▶ Earth?
 - 0.5 billion years for first life
 - 4 billion years for large life
 - 4.5 billion years for **us** to develop communication technology



Maybe assume 2 billion years as a minimum?

This gives a maximum mass of about 2 Suns.